Chapter and Topic	Lesson	Outcomes	Homework	
C1 Atomic structure	Atoms	<ul> <li>State the definition of an element</li> <li>Describe the basic structure of the periodic table</li> <li>Describe the basic structure of an atom</li> </ul>		
	History of the atom	<ul> <li>State how and why the atomic model has changed over time</li> <li>Explain that scientific theories are revised or replaced by new ones in the light of new evidence</li> </ul>	Draw a timeline to show the developments in the atomic model	
	Structure of the atom	<ul> <li>State the location, relative charge and relative mass of the protons, neutrons and electrons in an atom</li> <li>Describe what the atomic number and mass number of an atom represent</li> <li>Explain why atoms have no overall charge</li> </ul>		
	lons, atoms and isotopes	<ul> <li>Describe how to work out the number of protons, neutrons and electrons in an ion</li> <li>Define isotopes</li> </ul>	Identify the number of electrons, neutrons and protons in an atom	
	Electronic structures	<ul> <li>Describe how the electrons are arranged in an atom</li> <li>Present the electronic structures of the first 20 elements in the periodic table</li> </ul>	Draw out the electronic structure of Li, K and Mg	
	Chemical equations	<ul> <li>Describe what happens to the atoms in a chemical reaction</li> <li>Describe how the mass of reactants compares with the mass of the products</li> </ul>		

## Year 9 Chemistry SoW and HW plan Autumn 2

	•	Construct balanced symbol equations, including state symbols, to represent reactions	
Separating mixtures	•	State what a mixture is How to separate the components in a range of mixtures by <ul> <li>Filtration</li> <li>Crystallisation</li> <li>Simple distillation</li> </ul>	Kerboodle interactive quiz – separating mixtures
Fractional distillation and paper chromatography	•	Explain why fractional distillation is needed to separate some liquids Describe how fractional distillation works Describe how paper chromatography works	